



Deliverable 1.11

Report on optimized levulinic acid and ester route via Furfural

Demonstration of solvent and resin production from lignocellulosic biomass via the platform chemical levulinic acid

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About GreenSolRes

The need to establish economic and sustainable large-scale operations for the conversion of renewable resources to chemical building blocks is becoming increasingly urgent in the context of climate change and depleting fossil fuel reservoirs. Pathways for manufacturing of bio-based fuels and chemicals have been developed but most of them rely on sugar and starch crops for feedstock. GreenSolRes aims at a sustainable and competitive industrial production of the platform chemical levulinic acid (LVA) from non-food lignocellulosic biomass. Further, the conversion of LVA and LVA esters into industry relevant building blocks γ -valerolactone (GVL), 1-methyl-1,4-butanediol (MeBDO) and 2-methyltetrahydrofuran (2-MTHF) will take place by new catalytic methods developed during the course of this project. Finally, these chemicals will be upgraded to solvents and resin monomers for the production of high added value adhesives and consumer products. This project was started in September 2016.

Project Coordinator



Project Office



Consortium



About this document

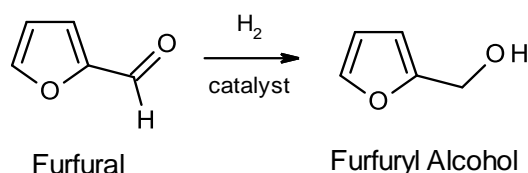
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Publishable Summary

The present report describes the work on the route from furfural via furfuryl alcohol to levulinic acid and alkyl levulinate respectively.

According to market reports, both furfural and furfuryl alcohol are existing commodities on an industrial scale of several 100 kilotons per year. An in-depth literature survey was conducted to assess the current state of the art for the reaction of furfural to furfuryl alcohol. Experimental confirmation was achieved using commercially available catalysts and furfural from European and Asian sources.



The transformation of furfuryl alcohol to levulinic acid and alkyl levulinate has been examined in more detail in discontinuous as well as continuous mode. It could be demonstrated that variation of reaction parameters like solvent, acid type, concentration and temperature have significant influence on the extent of side-product formation. With beneficial combination of parameters very high product selectivity was achieved. Results from batch experiments were successfully transferred to continuous process mode.

